Write your name here			
Surname		Other names	5
Pearson Edexcel International Advanced Level	Centre Number		Candidate Number
Physics Advanced Unit 5: Physics from	Creation to	o Colla	apse
Wednesday 21 June 2017 – Time: 1 hour 35 minutes	- Morning		Paper Reference WPH05/01
You do not need any other ma	aterials.		Total Marks

Instructions

- Use **black** ink or ball-point pen.
- **Fill in the boxes** at the top of this page with your name, centre number and candidate number.
- Answer **all** questions.
- Answer the questions in the spaces provided
 - there may be more space than you need.

Information

- The total mark for this paper is 80.
- The marks for **each** question are shown in brackets
 - use this as a guide as to how much time to spend on each question.
- Questions labelled with an asterisk (*) are ones where the quality of your written communication will be assessed
 - you should take particular care with your spelling, punctuation and grammar, as well as the clarity of expression, on these questions.
- The list of data, formulae and relationships is printed at the end of this booklet.
- Candidates may use a scientific calculator.

Advice

- Read each question carefully before you start to answer it.
- Try to answer every question.
- Check your answers if you have time at the end.

Turn over ▶



P48425A
©2017 Pearson Education Ltd.



SECTION A

Answer ALL questions.

For questions 1–10, in Section A, select one answer from A to D and put a cross in the box ⊠. If you change your mind, put a line through the box ₩ and then mark your new answer with a cross ⋈.

	mark your new answer with a cross ⋈.
1	Astronomers use standard candles to make measurements of distances to star clusters.
	A standard candle is an object of known
	■ A brightness.
	B distance.
	C luminosity.
	D size.
	(Total for Question $1 = 1$ mark)
2	Some mobile phones may be set to vibrate rather than ring. A mobile phone is placed on a wooden table and sets the table into oscillation as it vibrates. The vibration now sounds louder.
	This is an example of
	This is an example of
	This is an example of ☑ A free oscillation.
	This is an example of ■ A free oscillation. ■ B forced oscillation.
	This is an example of ■ A free oscillation. ■ B forced oscillation. ■ C resonance.

- 3 Radioactive decay is a random process.
 - 'Random' means that we cannot know
 - A how much energy will be released.
 - **B** the probability that a nucleus will decay per unit time.
 - C what type of radiation will be emitted.
 - **D** which nucleus will decay next.

(Total for Question 3 = 1 mark)

4 If hot liquid wax is left to cool, it will start to solidify at 45°C. The temperature of the wax will remain constant until all the wax has solidified.

Select the row in the table that correctly describes how molecular energy will change as the wax solidifies.

		Molecular kinetic energy	Molecular potential energy
X	A	decreases	decreases
X	В	decreases	stays constant
X	C	stays constant	decreases
X	D	stays constant	stays constant

(Total for Question 4 = 1 mark)

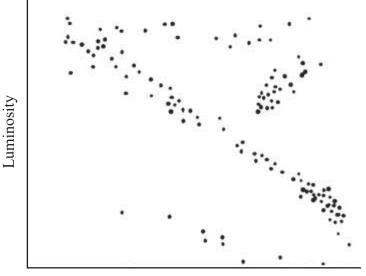
5 In a damped oscillation, a force acts to change the amplitude of oscillation.

Which of the following statements about the direction of this force is correct?

- A The force is in the opposite direction to the displacement.
- **B** The force is in the opposite direction to the velocity.
- C The force is in the same direction as the displacement.
- **D** The force is in the same direction as the velocity.

(Total for Question 5 = 1 mark)

6 The Hertzsprung-Russell diagram is a scatter graph of luminosity against temperature for stars.



Temperature

A star is located near to the top end of the main sequence.

Select the row in the table that correctly compares the mass and temperature of the star to those of the Sun.

		Mass	Temperature
X	A	larger	higher
X	В	larger	lower
X	C	smaller	higher
X	D	smaller	lower

(Total for Question 6 = 1 mark)

7 Stellar parallax is used by astronomers to determine the distances to stars.

Why can this method only be used for some stars?

- A The effect is too small for nearby stars.
- **B** The effect is too large for nearby stars.
- C The effect is too small for distant stars.
- **D** The effect is too large for distant stars.

(Total for Question 7 = 1 mark)

8 Nuclear fusion requires extreme conditions for the hydrogen fuel.

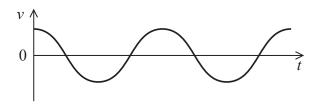
In order for any nuclei to fuse, which of the following is essential?

- A very high density
- B very high pressure
- C very high temperature
- D very large mass

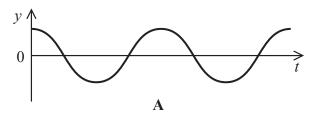
(Total for Question 8 = 1 mark)

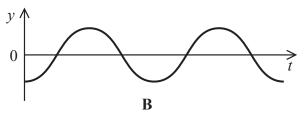
Questions 9 and 10 relate to the graph below.

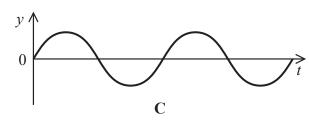
The graph shows how the velocity v varies with time t for an object undergoing simple harmonic motion.

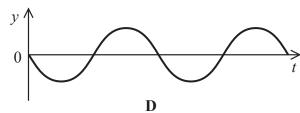


The following graphs show how other quantities for the object may vary with t over the same time period.









- 9 Select the graph which shows the variation of displacement from the equilibrium position with time.
 - \mathbf{X} \mathbf{A}
 - \mathbf{R}
 - \mathbf{K} C
 - \square D

(Total for Question 9 = 1 mark)

- 10 Select the graph which shows the variation of acceleration with time.
 - \mathbf{X} \mathbf{A}
 - \blacksquare B
 - \square C
 - \square D

(Total for Question 10 = 1 mark)

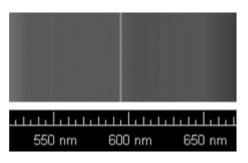
TOTAL FOR SECTION A = 10 MARKS

SECTION B

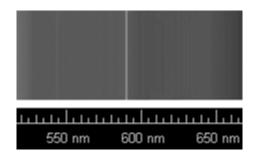
Answer ALL questions in the spaces provided.

11 The two images show a small part of the spectrum produced by helium. One image is a spectrum produced in a laboratory and the other is a spectrum produced by the light from a star.

Both images contain a bright yellow line. In the spectrum produced by light from the star, the yellow line is shifted in wavelength.



spectrum produced in a laboratory

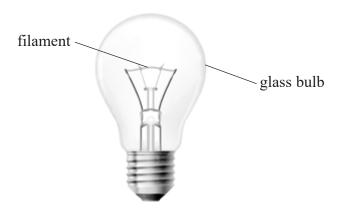


spectrum produced by light from a star

Calculate the magnitude of the velocity of the star relative to the Earth.	(3)
Velocity of sta	r =
(Total for Question 1	11 = 3 marks)



12 Some light bulbs consist of a metal filament inside a glass bulb. The bulb may be filled with a mixture of gases to reduce vaporisation of the filament.



(a) The glass bulb contains a mixture of krypton gas and xenon gas at room temperature. The mean squared speed of the krypton molecules is 8.72×10^4 m² s⁻².

Calculate the mean squared speed of the xenon molecules.

mass of 1 molecule of krypton = 1.39×10^{-25} kg mass of 1 molecule of xenon = 2.18×10^{-25} kg

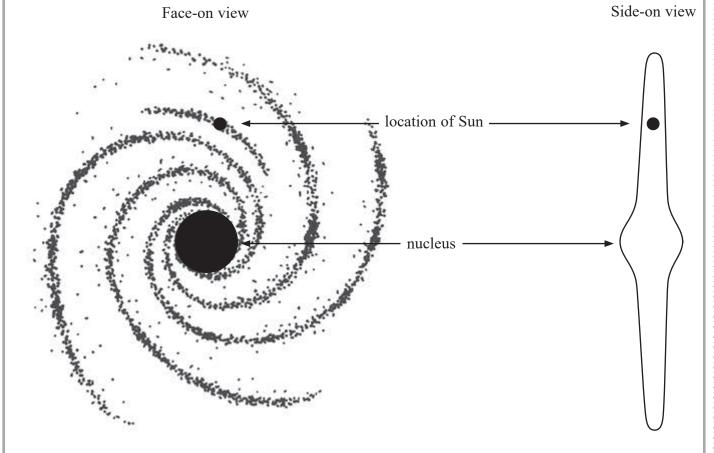
	7	1
U	L	J
1		/

Mean squared speed = $m^2 s^{-2}$



*(b)	When a current flows through the filament, the gas in the bulb heats up and the pressure exerted by the gas increases.	
	Explain, including ideas of momentum, why the pressure exerted by the gas increases	S .
		(4)
	(Total for Question 12 = 6 mar	rks)
	(100miloi Question 12 0 mai	~ /

13 The Sun is one of about 10¹¹ stars in the Milky Way galaxy. The vast majority of stars in the Milky Way are observed to be within its nucleus. The Sun lies on the Sagittarius arm of the Milky Way, about 27 000 light years from the centre of the galaxy, as shown.



- (a) The galaxy is rotating about its centre and astronomers estimate that the Sun makes one complete revolution every 240 million years.
 - (i) Show that the angular velocity of the Sun $\omega_{\rm Sun}$, about the centre of the galaxy, is about $8\times 10^{-16}~{\rm rad~s^{-1}}$.

1 year =
$$3.15 \times 10^7 \, \text{s}$$



		the centre of the	(2)
(''') II 41		£ £ (') 4-	11-4-
(iii) Use the expression from (ii), together a value for the angular velocity $\omega_{\rm star}$	r with the value of a star similar to	of $\omega_{_{\mathrm{Sun}}}$ from (1), to the Sun, but loc	ated at the
edge of the Milky Way.			
diameter of galaxy $\approx 100~000$ light y	ears		(2)
			(-)
		$\omega_{ m star} =$	
) In fact, stars similar to the Sun, but furth	ner away from the	centre of the gala	axv. are
observed to have angular velocities that			my, are
Explain what astronomers can conclude	from these observ	ations.	
			(2)



14 The photograph shows a dancing "hula girl" toy. The toy uses energy from the Sun to make the girl dance. When the solar cell is illuminated the girl's arms move with simple harmonic motion.



((a)	State th	e conditions	for an o	biect to	move with	simple	harmonic	motion
•	a)	State III	e conamons	101 all (JUJECI IU	move with	SIIIIbic	Harmonic	monon.

(2)

(b) (i) The time t for 50 oscillations of the arms is measured three times and the values obtained are recorded in the table.

<i>t</i> ₁ /s	<i>t</i> ₂ /s	<i>t</i> ₃ /s
18.9	19.2	19.1

Show that the arms oscillate with a frequency of about 2.6 Hz.

(3)

(ii)	From the top to the bottom of the movement, the hands travel a distance of 0.75 c	em.
	Calculate the maximum speed of the hands.	(3)
	Maximum speed of hands =	
	(Total for Question 14 = 8 mar	rks)

15	Basketballs are usually made from a leather composite and are filled with air. A standard basketball has a total mass of 0.620 kg and a volume of 8.18×10^{-3} m ³ .										
	(a) A basketball is filled with air at a pressure of 1.55×10^5 Pa	and at a temperature of 20 °C.									
	Calculate the number of air molecules inside the basketball.	(3)									
	Number	of air molecules =									
	(b) A standard basketball rebounds to a height of between 1.40 dropped onto a hard surface from a height of 1.80 m.	m and 1.60 m when									
(i) Show that the maximum allowable decrease in kinetic energy during the bounce is about 2.4 J.											
	about 2.43.	(3)									

	Calculate the number of times a standard basketball must be dropped from 1.80 to increase the temperature of the basketball and the air inside by 0.5 °C.	m
		(3)
	·	
	Number of times basketball must be dropped =	
····		
(111)	State one assumption that you had to make to calculate the number of times that the basketball had to be dropped.	
		(1)

ealed in plastic packages and then passed in front of a radioact	iating them. The instruments a ive isotope.
a) By considering relevant properties of each type of radiation or γ radiation would be most appropriate for this method.	, determine whether α , β ,
,	(4)
5.27 years. A sample of this isotope contains 5.02×10^{13} ur	nstable nuclei when it is new.
 b) A radioactive isotope commonly used as a source for the rad 5.27 years. A sample of this isotope contains 5.02 × 10¹³ ur (i) Show that the activity of the sample when it is new is al 1 year = 3.15 × 10⁷ s 	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.
5.27 years. A sample of this isotope contains 5.02×10^{13} ur (i) Show that the activity of the sample when it is new is all	nstable nuclei when it is new.

Calculate the time, in years, until the activity of the sample decre	
	(3)
Time	=
	=
	=
A technician is checking the activity of a radioactive source.	=(2)
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	
A technician is checking the activity of a radioactive source.	

- 17 Americium-241(Am) is an artificially produced radioactive isotope which was commonly used in smoke detectors. It decays through emission of an α -particle to neptunium (Np).
 - (a) (i) Complete the nuclear equation which represents the decay of americium.

(2)

$$^{241}_{95}Am \rightarrow Np + \alpha$$

*(ii) Explain why most of the energy released in the decay becomes kinetic energy of the α -particle.

(3)

(b) The table gives masses of an americium nucleus and its constituent particles.

	Mass / u
¹ p	1.00728
¹n	1.00866
²⁴¹ Am	241.00471

Calculate the binding energy p	r nucleon, in J, o	f americium-241.
--------------------------------	--------------------	------------------

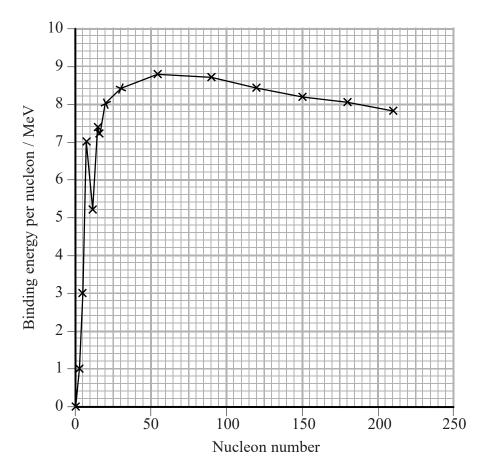
(4)

|
 |
|------|------|------|------|------|------|------|------|------|------|------|------|------|------|------|------|
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| |
 | | | |
 | | |
 |
 | | | |
 | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
| | | | | | | | | | | | | | | | |
|
 |

Binding energy per nucleon = _____ J



(c) The graph shows how the binding energy per nucleon varies with nucleon number for a range of nuclides.



Use the graph to explain why the fission of nuclei with large numbers of nucleons releases large amounts of energy.

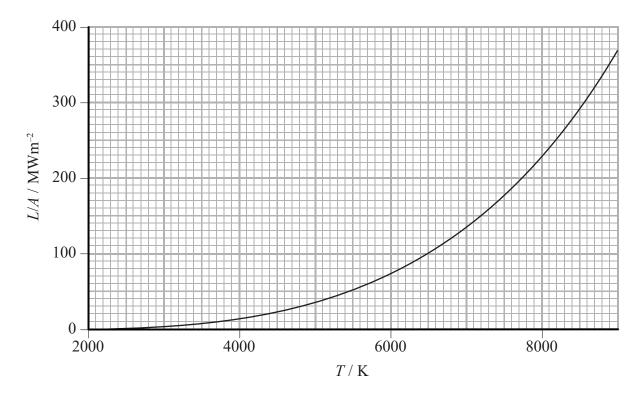
(3)

(Total for Question 17 = 12 marks)

BLANK PAGE



18 A black body of surface area A has luminosity L. The graph shows how the radiated flux L/A varies with temperature T.



(a) (i) Using data from the graph show that the luminosity of the Sun is about $4\times10^{26}\,\mathrm{W}.$

surface temperature of the Sun = 5900 K radius of the Sun = 6.96×10^8 m

(3)

	Calculate the radiation flux F from the Sun at the top of the Earth's atmosphere. distance from Sun to Earth = 1.50×10^{11} m	(2)
I	Suggest why the total rate at which energy from the Sun arrives at the top of the Earth's atmosphere is much less than $F \times A$, where A is the area of the spherical surface representing the top of the atmosphere.	(2)



is about 5×10^{-7} m.	which peak energy emission occurs for the	Sun
		(2)
(ii) The visible region of the electron about 400 nm to a wavelength o	magnetic spectrum extends from a wavelengt about 700 nm.	gth of
Suggest why light from the Sun	is white rather than yellow.	
		(2)

TOTAL FOR SECTION B = 70 MARKS
TOTAL FOR PAPER = 80 MARKS

List of data, formulae and relationships

Acceleration of free fall	$g = 9.81 \text{ m s}^{-2}$	(close to Earth's surface)
---------------------------	-----------------------------	----------------------------

Boltzmann constant
$$k = 1.38 \times 10^{-23} \text{ J K}^{-1}$$

Coulomb's law constant
$$k = 1/4\pi\epsilon_0$$

$$= 8.99 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$$

Electron charge
$$e = -1.60 \times 10^{-19} \text{ C}$$

Electron mass
$$m_e = 9.11 \times 10^{-31} \text{kg}$$

Electronvolt
$$1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$$

Gravitational constant
$$G = 6.67 \times 10^{-11} \text{ N m}^2 \text{ kg}^{-2}$$

Gravitational field strength
$$g = 9.81 \text{ N kg}^{-1}$$
 (close to Earth's surface)

Permittivity of free space
$$\varepsilon_0 = 8.85 \times 10^{-12} \text{ F m}^{-1}$$

Planck constant
$$h = 6.63 \times 10^{-34} \,\mathrm{J s}$$

Proton mass
$$m_{\rm p} = 1.67 \times 10^{-27} \, \text{kg}$$

Speed of light in a vacuum
$$c = 3.00 \times 10^8 \,\mathrm{m \, s^{-1}}$$

Stefan-Boltzmann constant
$$\sigma = 5.67 \times 10^{-8} \ W \ m^{-2} \ K^{-4}$$

Unified atomic mass unit
$$u = 1.66 \times 10^{-27} \text{ kg}$$

Unit 1

Mechanics

Kinematic equations of motion
$$v = u + at$$

$$s = ut + \frac{1}{2}at^2$$
$$v^2 = u^2 + 2as$$

Forces
$$\Sigma F = ma$$

$$g = F/m$$

$$W = mg$$

Work and energy
$$\Delta W = F \Delta s$$

$$E_{\rm k} = \frac{1}{2}mv^2$$

$\Delta \hat{E}_{ m grav}^{}=mg\Delta h$

Materials

Stokes' law
$$F = 6\pi \eta r v$$

Hooke's law
$$F = k\Delta x$$

Density
$$\rho = m/V$$

Pressure
$$p = F/A$$

Young modulus
$$E = \sigma/\varepsilon$$
 where

Stress
$$\sigma = F/A$$

Strain $\varepsilon = \Delta x/x$

Elastic strain energy
$$E_{\rm el} = \frac{1}{2}F\Delta x$$



Unit 2

Waves

Wave speed $v = f\lambda$

Refractive index $\mu_2 = \sin i / \sin r = v_1 / v_2$

Electricity

Potential difference V = W/Q

Resistance R = V/I

Electrical power, energy and P = VI efficiency $P = I^2R$

 $P = V^2/R$

W = VIt

% efficiency = $\frac{\text{useful energy output}}{\text{total energy input}} \times 100$

% efficiency = $\frac{\text{useful power output}}{\text{total power input}} \times 100$

Resistivity $R = \rho l/A$

Current $I = \Delta Q/\Delta t$

I = nqvA

Resistors in series $R = R_1 + R_2 + R_3$

Resistors in parallel $\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3}$

Quantum physics

Photon model E = hf

Einstein's photoelectric $hf = \phi + \frac{1}{2}mv_{\text{max}}^2$

equation



Unit 4

Mechanics

Momentum p = mv

Kinetic energy of a

non-relativistic particle $E_k = p^2/2m$

Motion in a circle $v = \omega r$

 $T=2\pi/\omega$

 $F = ma = mv^2/r$

 $a = v^2/r$

 $a = r\omega^2$

Fields

Coulomb's law $F = kQ_1Q_2/r^2$ where $k = 1/4\pi\varepsilon_0$

Electric field E = F/Q

 $E = kQ/r^2$

E = V/d

Capacitance C = Q/V

Energy stored in capacitor $W = \frac{1}{2}QV$

Capacitor discharge $Q = Q_0 e^{-t/RC}$

In a magnetic field $F = BIl \sin \theta$

 $F = Bqv \sin \theta$

r = p/BQ

Faraday's and Lenz's laws $\varepsilon = -d(N\phi)/dt$

Particle physics

Mass-energy $\Delta E = c^2 \Delta m$

de Broglie wavelength $\lambda = h/p$



Unit 5

Energy and matter

Heating $\Delta E = mc\Delta\theta$

Molecular kinetic theory $\frac{1}{2}m\langle c^2 \rangle = \frac{3}{2}kT$

Ideal gas equation pV = NkT

Nuclear Physics

Radioactive decay $dN/dt = -\lambda N$

 $\lambda = \ln 2/t_{_{1/2}}$

 $N = N_0 e^{-\lambda t}$

Mechanics

Simple harmonic motion $a = -\omega^2 x$

 $a = -A\omega^2 \cos \omega t$ $v = -A\omega \sin \omega t$ $x = A \cos \omega t$ $T = 1/f = 2\pi/\omega$

Gravitational force $F = Gm_1m_2/r^2$

Observing the universe

Radiant energy flux $F = L/4\pi d^2$

Stefan-Boltzmann law $L = \sigma T^4 A$

 $L = 4\pi r^2 \sigma T^4$

Wien's law $\lambda_{\max} T = 2.898 \times 10^{-3} \text{ m K}$

Redshift of electromagnetic

radiation $z = \Delta \lambda / \lambda \approx \Delta f / f \approx v / c$

Cosmological expansion $v = H_0 d$